

Chemistry 125/126

Welcome
What? Why? How?

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Chemistry 125 vs 126?

CHEM.125/126

- Co-requisites with identical work and grades.
- One credit each for a total of two credits.
- Credit for TWO inorganic lab courses.

WAIT LISTS AND OVERRIDES

- Report to 1500 chem. (administrative office)
- Contact Debra Buck; debbuck@umich.edu

Course Information

- An independent introductory gen chem course with its own lecture, lab, and discussion
- Labs and discussion on a given topic occur after pre-lab lecture.
- Instructors for lab and discussion are GSIs.

Course Format

- Pre-lab lecture (1 hour) in 1800 chemistry
- Discussion (1 or 0 hour) in lab room
- Laboratory (2 or 3 hours) in lab room

- Labs = 3 hours if discussion is NOT held; labs = two hours when discussion is held
- Discussions occur AFTER the lab experiment is done

Required Materials

Collaborative Investigations in Chemistry,
Nancy Konigsberg Kerner and James Penner-Hahn,
Hayden McNeil Publishing, Inc., Fall 2010 edition

Supplies
Lab Marking pens



Web Sites

Ctools

- Course Information
- Resources: Lectures, Schedule, Exams, Review notes
- iTunes: Lecture Podcasts, Review Podcasts, Video Resources
- Lab data: Link to course website with evolving lab data

<http://www.umich.edu/~chem125>

Lab Safety

- Contact Lenses may NOT be worn in lab.
- Goggles and aprons must be worn in lab.



Lab Safety

- No open shoes -- sandals, thongs, etc
- Clothing should adequately cover your body – i.e. no shorts etc.
- Follow safety rules (pp.18-19, manual) at all times



Special Needs

Special Safety Problems?

- Richard Giszczak; 1608 chem.
- richg@umich.edu



Special Needs or Concerns?

- Nancy Kerner; 3541 chem.
- nkerner@umich.edu
- Office hours:
Tuesdays, 3-4 pm, Thursdays, 1-2 pm

Chemistry 125/126 Grading

TOTAL course points	500 points
Lab and discussion (6)	300 points*
Exams (2)	150 pts
GSI/peer points	50 points
* See Course Information for lab and discussion point details	

- Points reward individual and team **efforts**.

Team % points	54 %
Individual % points	46 %

Chemistry 125/126 Grading

- Grades based on explicit point criteria to establish mastery; there is NO grading on a curve

Fall 2010 guaranteed course letter grades:

At least an A-	450 points
At least a B-	400 points
At least a C-	350 points
At least a D-	300 points

- Point grade cutoffs guaranteed.
- Points needed for a particular letter grade will not be increased *but may be lowered* if some aspect of grading is not equitable to prior terms.

Lab Performance and Points

- Labs must be performed to earn points
- Miss two labs? Contact the course coordinator for permission to stay in the course.
- Makeup may be arranged informally with home GSI to be performed in his/her other section
- Formal makeup directives are in the back of your lab manual
- Participation in discussion is required to earn discussion points

Lab Makeups

- When arranging a time/day for a makeup check the schedule of labs:

	Mon	Tues	Wed	Thurs	Fri
8-11				✓	✓
11-2	✓	✓	✓		✓
2-5		✓	✓	✓	

- The lab week runs from Wednesday thru Tuesday
- You will perform an individual (rather than team) makeup if your makeup occurs outside the lab week

Earning Maximum Points

Team Reports

- The form in the manual is merely an outline.
- For maximum points respond to all questions in the experiment
- Refer to lab data to support conclusions**
- Study the "Team Report Tips" in the manual
- Study the experiment grading rubrics

Discussion

- Study the Discussion Grading Rubrics

Exams

- There is no final exam!
- There are **two hourly exams** (75 pts each):
 - Tuesday, November 9, 6:15 - 7:45 pm
 - Monday, December 13, 6:15 - 7:45 pm
- There are **alternate exams on the same days** for students with legal conflicts or students needing extended time



Session One (September 8 - 14)

- Introductions**
- Team Assignment Survey Form**, manual, page 4

Break

- Team assignments**
- Check-In**
- Team Task Exercise**, pages 5- 7
- Team Task Schedule**, page 8
- Periodic Table Scavenger Hunt**, pages 12-15
- Safety and Scavenger Hunt**, pages 18 – 20

Pre-lab Prep and Schedule

(Manual, pages 239 - 240)

Experiment Topics	Pre-lab lecture	Pre-lab Reading Pages	Lab Points
Check-in	9/7 (300 sections)		
Safety Hunt	9/7 (200 sections)		
Team Task	9/9 (100 sections)	1-21	
Periodic Table Hunt		230-243	

Pre-lab Prep and Schedule

Experiment Topics	Pre-lab Lecture	Pre-lab Reading Pages	Lab Points
Experiment 1: Precipitation and Water Purity	9/14 (300 sections) 9/14 (200 sections) 9/16 (100 sections)	22-53 190-191 208-211	35

The Pre-lab report (page 38) for experiment 1:

- requires work on the Internet
- is due at the start of experiment 1.

Chem.125/126 Topics and Learning

"To develop learning competence students must understand facts and ideas in the context of a conceptual framework"

— Bransford, Brown, & Cocking, Eds.

How People Learn: Brain, Mind, Experience, and School.

- The **chem.125/126 framework** is the Periodic Table.



Chem.125/126 Topics and Learning

"To develop learning competence students must understand facts and ideas in the context of a conceptual framework"

- The chem.125/126 theme song, **"Structure and Property Relationships"**



Chem. 125/126 Goals and Methods

Student goals and background?

- Do you intend to be a chemist?

- Students do NOT intend to be chemists!



CHEMISTRY 125/126 Methods

Methods fueled by concern about what non-chemist students can do with the skills they learn later on in life and student learning research.

- **Develop life long skills**
Data analysis, team work, presentation skills...)
- **Understand core concepts**
Emphasize process rather than content or memorization



Boyer Commission Report: Carnegie Foundation for the Advancement of Teaching (1998 & 2004)

"Many undergraduates graduate without knowing how to think logically, write clearly, or speak coherently"

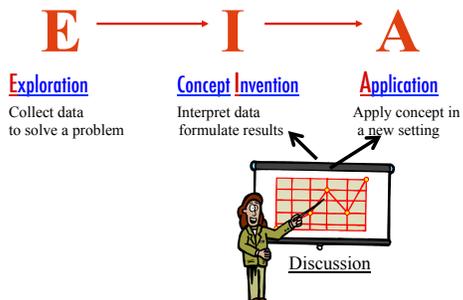
Alter classroom methods:

Traditional → Inquiry
Individual → Teamwork
Teacher-centered → Student-centered
→ Incorporate technology

Why Inquiry-based learning ?

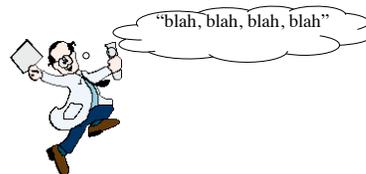
- Students construct their own understanding of **concepts**, rather than simply being told information
- Students develop **stronger critical thinking skills**
- The approach **exposes students to the process of science**

Chem.125/126: Guided Inquiry Labs



Why Student-Centered Course?

- **Teacher (rather than the student) becomes highly skilled in a teacher-centered class** - Teacher (rather than the student) writes, speaks, consults, organizes, and solves problems.



Why Teamwork?

- **Students differ in thinking and processing skills.**



- **Social interactions** reveal differences and result in learning by restructuring of students knowledge.

Chem.125/126: Why Teamwork?

- **Learning is a social activity**
- **Social interactions can help students build their knowledge base, monitor their progress, and clarify their understanding (and misunderstanding)**

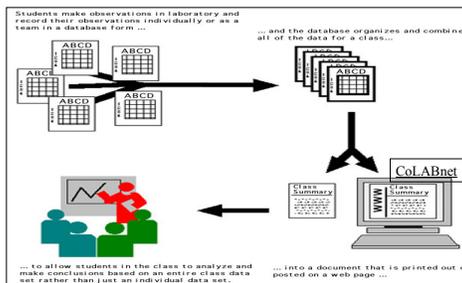


Lab Methods

- **Collaborative Team research.**
- **Teams collect data for different (rather than identical) samples and/or conditions.***
- **Technology assisted data collection and analysis.**

* See page 232 of the lab manual for team experiment and discussion question assignments.

CoLABnet



- **Data is available in lab and on the course website:**
<http://www.umich.edu/~chem125>.

Discussion Methods

- Teams solve assigned problems in lab.*
- Teams orally present results in discussion.
- Questions address critical thinking skills of invention and application

* See p.232 of lab manual for team assigned discussion questions.



Student Success in chem.125/126

The inquiry format and team work does NOT

- insure an A in the course
- insure student learning and understanding.



YOU (the student) need to make tactical decisions!

Student Success in Chem.125/126

- Choice of tactics must be appropriate.



Student Success in Chem.125/126

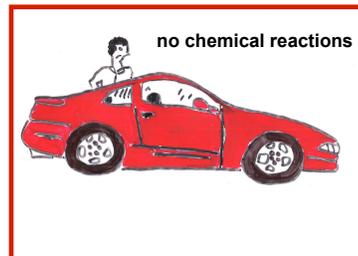
- Is NOT dependent on prior knowledge
- Is dependent on your individual efforts!
 - Come prepared to lecture, lab, and discussion
- Invest effort in your team work and social interactions
 - think out loud with teammates
 - instruct one another as to how solutions are derived
 - collaborate; don't split up work!
- Use available support and resources
 - use Ctools resources, office hours...
 - Study exams now!



Where Would We Be Without Chemistry?



Where Would We Be Without Chemistry?



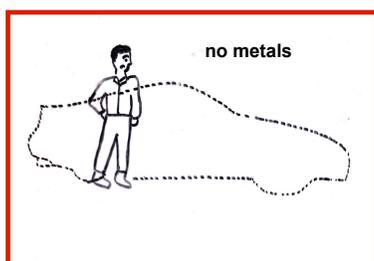
Where Would We Be Without Chemistry?



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Where Would We Be Without Chemistry?



Where Would We Be Without Chemistry?



Where Would We Be Without Chemistry?

150 kg male

H ₂ O	_____	50.1 kg
C	_____	12.6 kg
N	_____	1.8 kg
Ca	_____	1.7 kg
P	_____	.68 kg
K	_____	.25 kg
Na, Mg, Fe, etc	_____	.32 kg

Where Would We Be Without Chemistry?

No you



